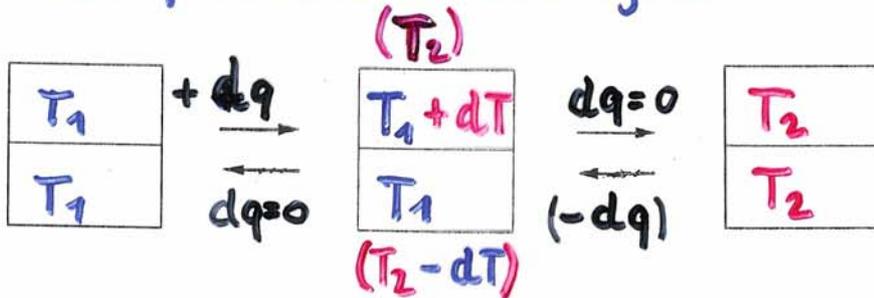


Reversible Prozesse: Nahe am Gleichgewicht

Temperatur - Gleichgewicht



Druckgleichgewicht: Reversible, isotherme Expansion (Kompression)

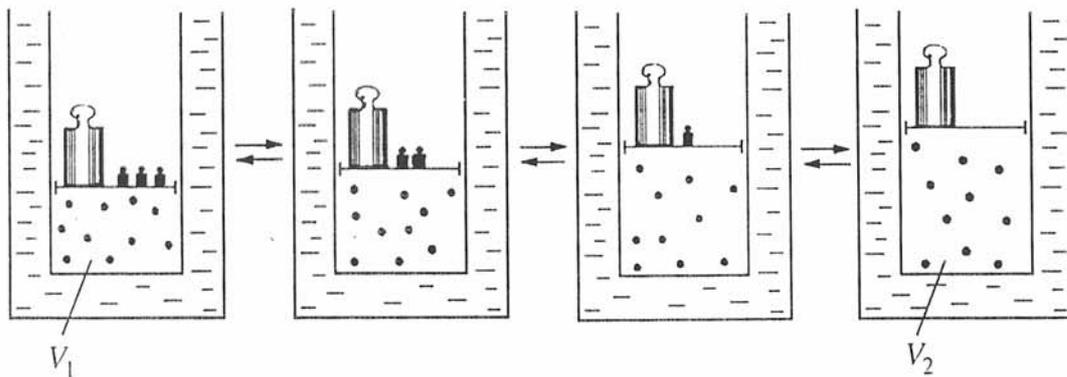
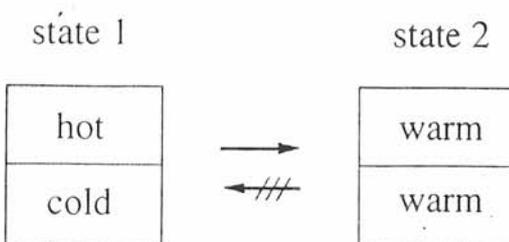
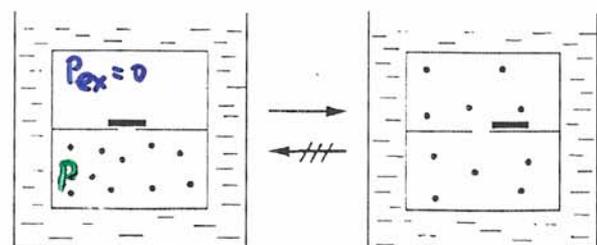


Figure 14.3 Reversible isothermal expansion of a gas (decrease of the pressure by stepwise removal of tiny weights) from volume V_1 to volume V_2 .

Irreversible Prozesse: Fernab des Gleichgewichts



Temperaturausgleich



Expansion ins Vakuum

$$P_{ex} = 0$$